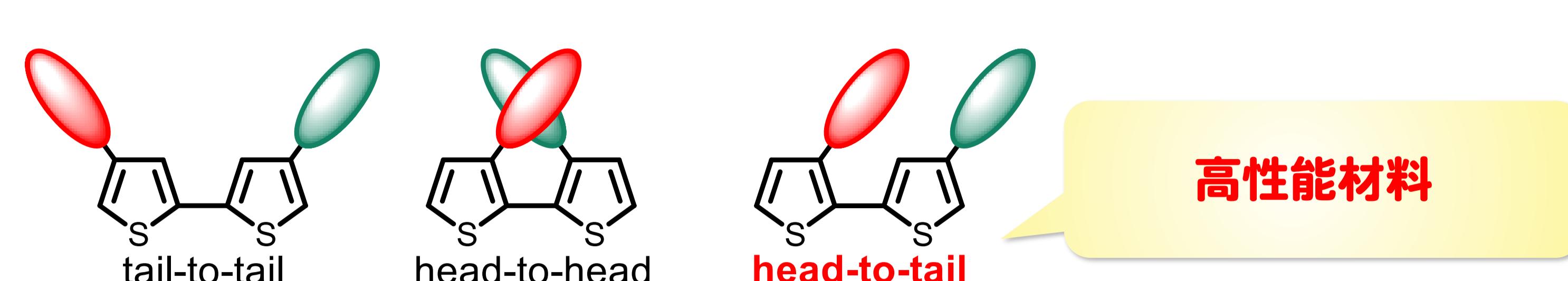


Background

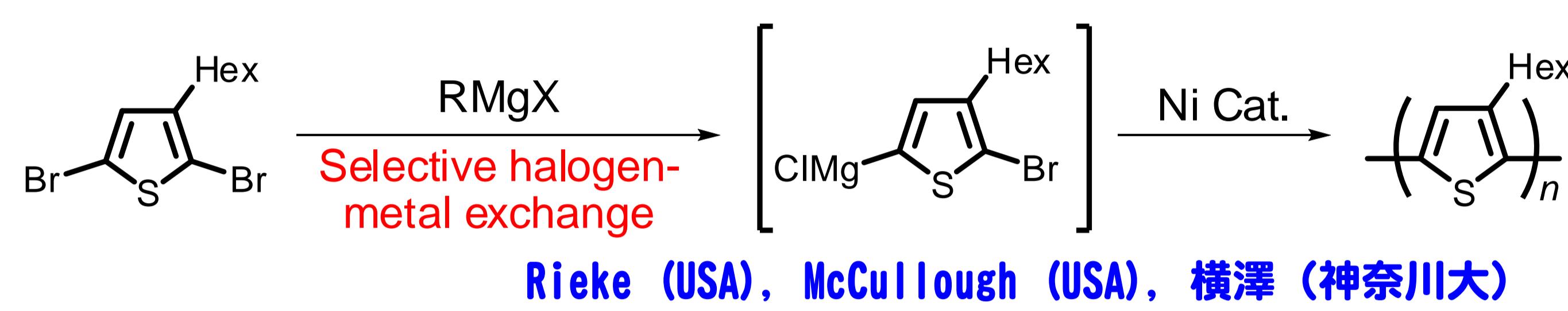
◇ Poly(3-hexylthiophene) (P3HT)

有機半導体
有機LED
有機薄膜太陽電池



高性能材料

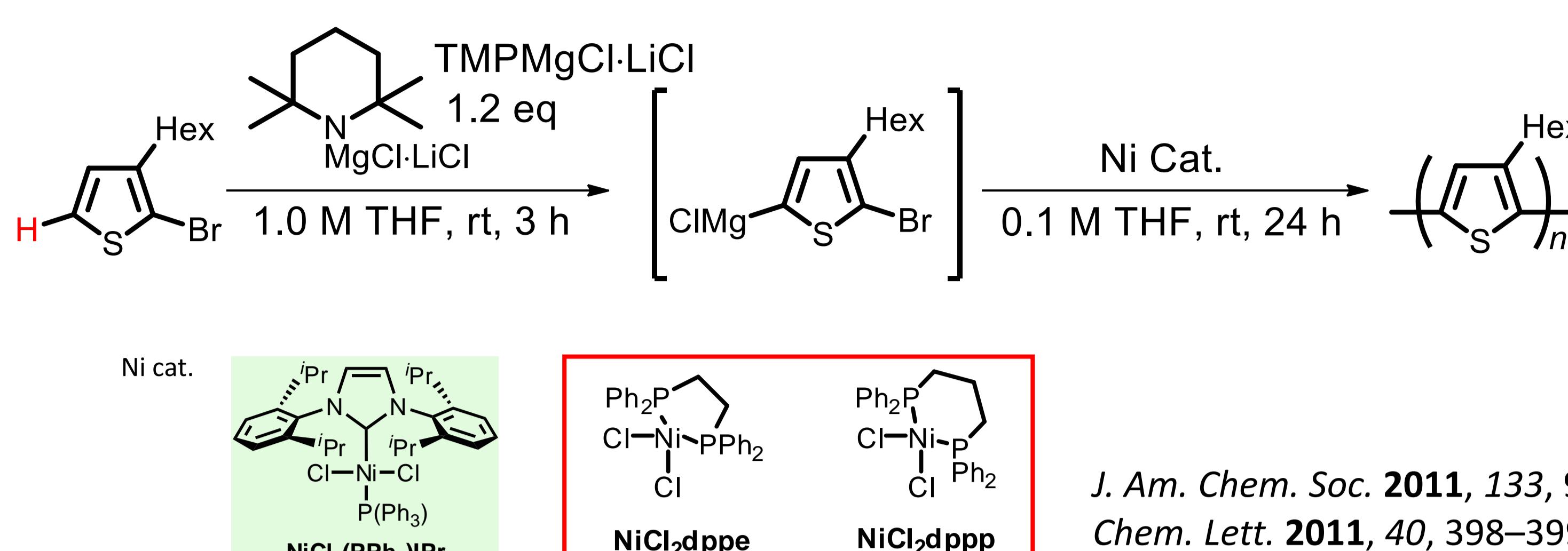
◆ 一般的なポリチオフェンの合成法 GRIM法



Rieke (USA), McCullough (USA), 横澤（神奈川大）

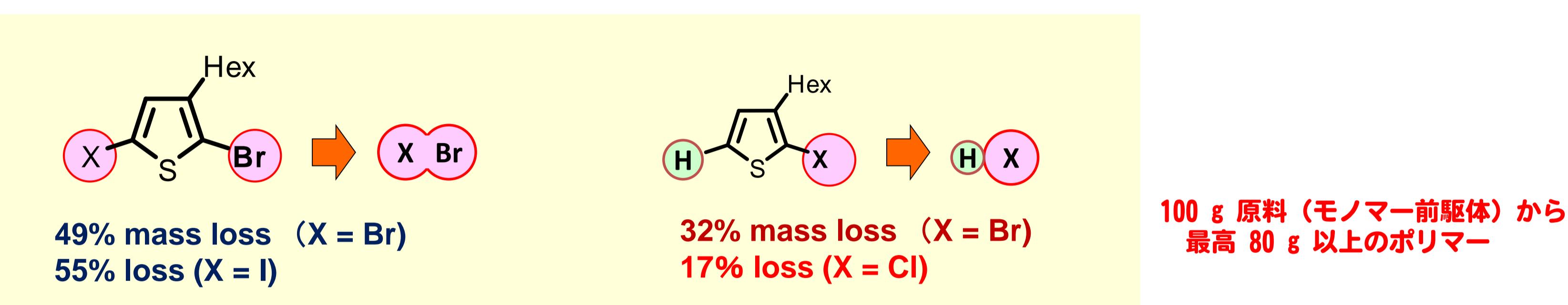
Osaka, I.; McCullough, R. D. Acc. Chem. Res. 2008, 41, 1202.

◇ 当研究グループ開発のポリチオフェン合成 C-Hカップリング重合



J. Am. Chem. Soc. 2011, 133, 9700–9703.
Chem. Lett. 2011, 40, 398–399.

◆ ポリチオフェン合成の原子効率

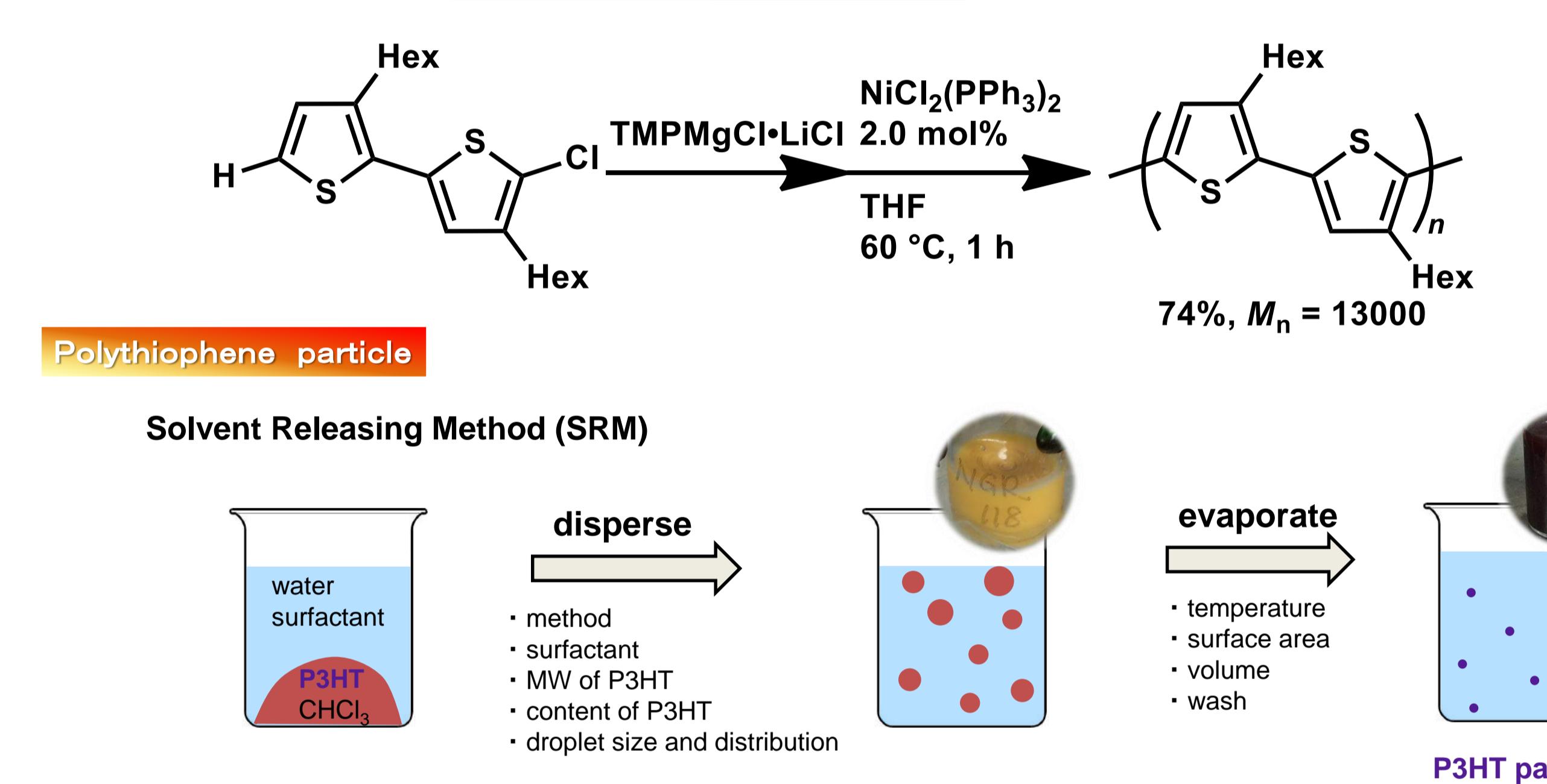


100 g 原料（モノマー前駆体）から
最高 80 g 以上のポリマー

Results & Discussion

◆ This work

Polymerization of bithiophene as a monomer

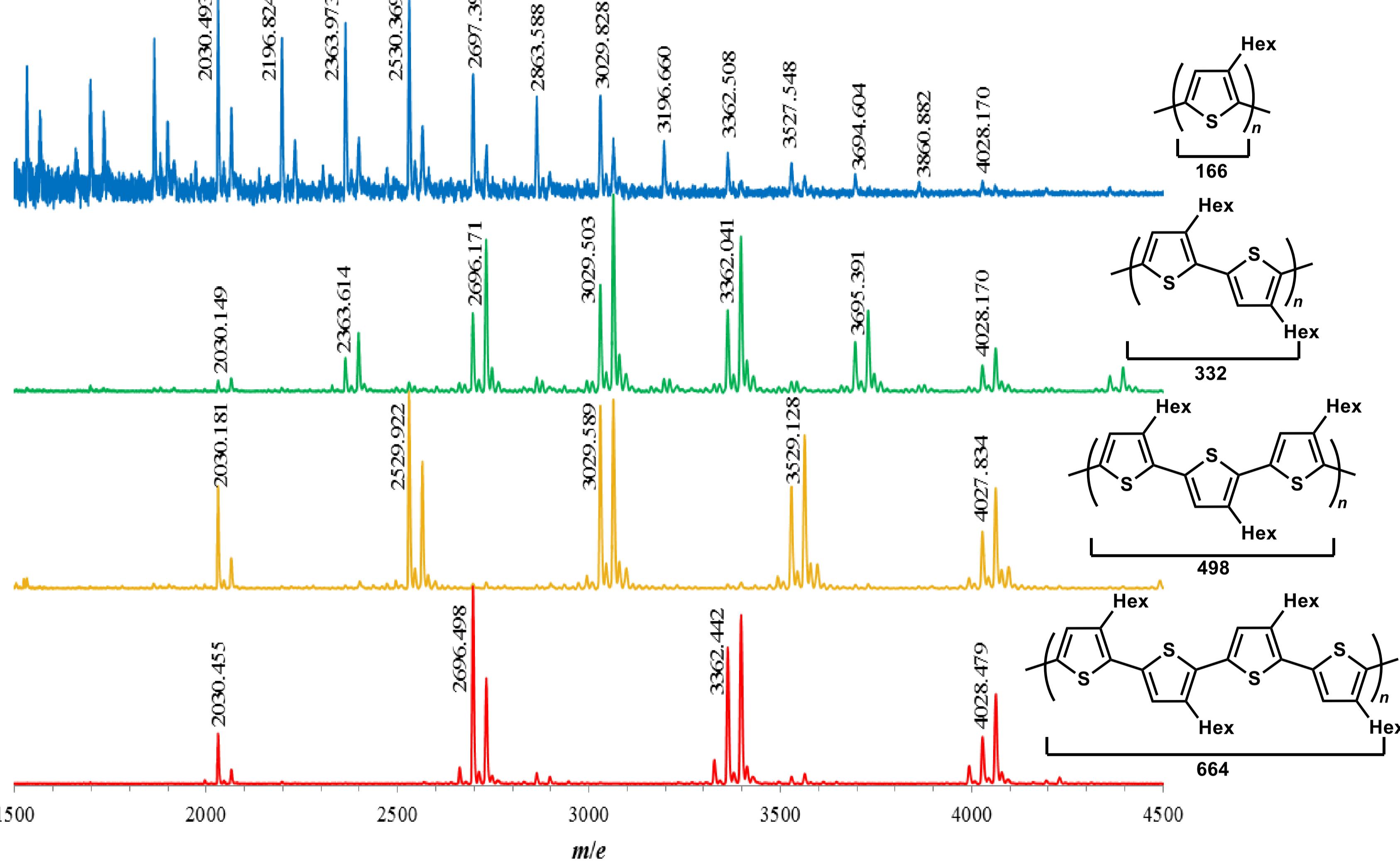


Polymerization of bithiophene

entry	catalyst	ligand	yield [%]	$M_n \times 10^{-3}$	M_w/M_n
1	$\text{NiCl}_2(\text{PPh}_3)_2$		74	13.0	1.67
2	$\text{NiCl}_2(\text{PPh}_3)_2$ ^{a)}		69	13.0	1.48
3	NiCl_2dppp		81	17.2	1.48
4	$\text{NiCl}_2(\text{PPh}_3)\text{IPr}$		84	39.0	1.67
5	$\text{NiCl}_2(\text{dme})$	2.5 $\text{P}(\text{o-tol})_3$	2	8.5	1.30
6	$\text{NiCl}_2(\text{dme})$	2.5 $\text{P}(\text{OEt})_3$	40	17.7	1.75
7	$\text{NiCl}_2(\text{dme})$	BPY	<1	6.9	1.23
8	$\text{Ni}(\text{acac})_2$	5 PPh_3	70	14.7	1.74
9	$\text{PdCl}_2(\text{PPh}_3)_2$		<1	--	--
10	PdCl_2dppe		<1	8.9	1.49
11	Pd-PEPPI-SIPr		28	5.7	1.15

^a The polymerization was carried out at r.t. for 3 h.

◆ MALDI-TOF mass spectra of poly(3-hexylthiophene)



◆ Preparation of P3HT particle

Preliminary result

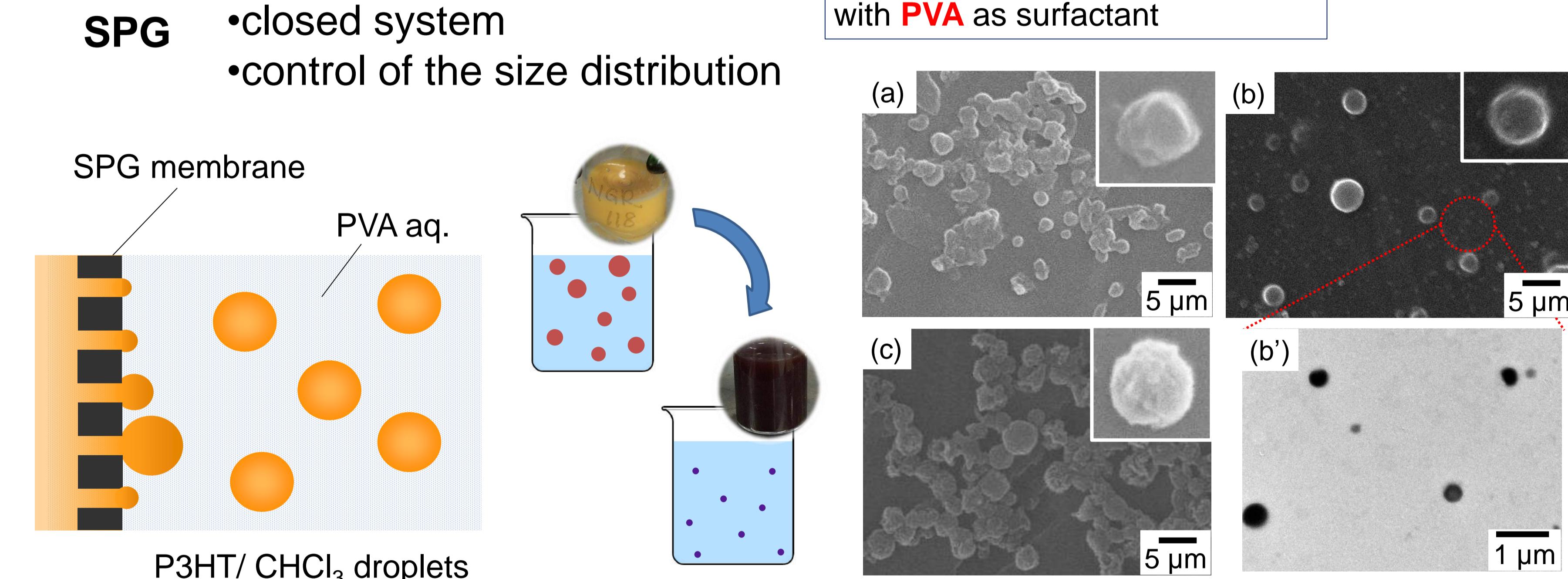
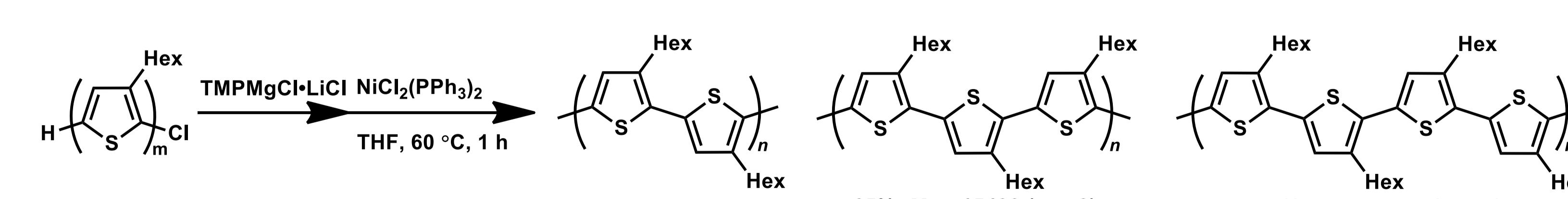


Fig. SEM (a-c) and TEM (b') images of P3HT particles prepared by solvent released method from the droplets of 0.5% P3HT chloroform solution with PVA. MW of P3HT: (a) 7,000, (b) 20,000, (c) 50,000.

Summary

◆ Ubiquitous and less reactive catalyst, $\text{NiCl}_2(\text{PPh}_3)_2$, effectively catalyzes polymerization of thiophene oligomers.



◆ Polythiophene particles

